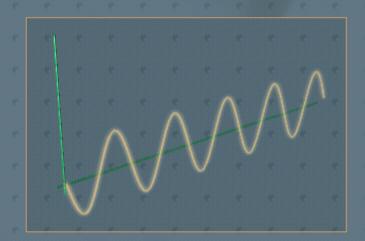


Light may be understood as propagating waves. Sound may also be described as propagating waves.

However, light waves are [electromagnetic] waves, [sound] waves are not.

All electromagnetic waves propagate at the [speed of light].



PREVIOUS

NEXT

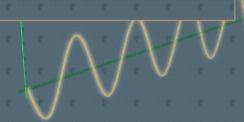
Light may be understood as propagating

waves. Sound may also b propagating waves.

In contrast with sound waves, electromagnetic waves do not need a medium in which to travel. Electromagnetic waves can even travel in vacuum.

However, light waves are waves, [sound] waves are not.

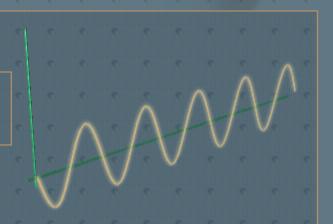
All electromagnetic waves propagate at the [speed of light].



Light may be understood as propagating waves. Sound may also be described as propagating waves.

Sound waves require a medium in which to travel. Air or water are typical examples. In air, sound travels at a velocity of around 330 meters per second.

All electromagnetic waves propagate at the [speed of light].



PREVIOUS

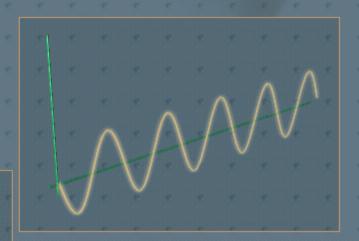
NEXT

Light may be understood as propagating waves. Sound may also be described as propagating waves.

However, light waves are [electromagnetic] waves, [sound] waves are not.

All electromagnetic waves propagate at the

Light travels in vacuum at almost 300 000 kilometers per second. This is almost 1 million times faster than the velocity of sound traveling in air.



PREVIOUS

NEYT

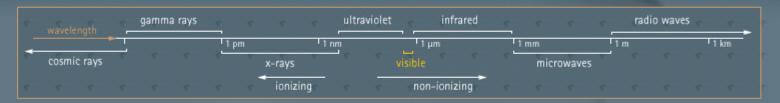
Light is a part of the electromagnetic spectrum.



Waves differ from each other by their [wavelength].

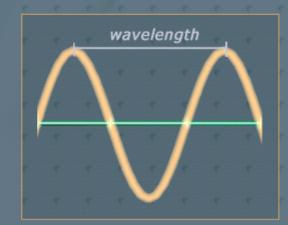
Electromagnetic waves of all wavelengths make up the electromagnetic spectrum.

Light is a part of the electromagnetic spectrum.



The wavelength is defined as the distance between two peaks of a wave. The wavelength is frequently symbolized by the greek letter lambda (λ) and ranges over many [orders of magnitude].

electromagnetic spectrum.



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lambda (λ.) and ranges

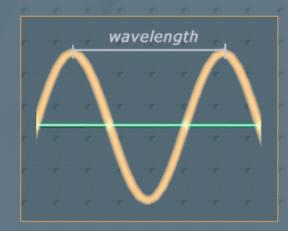
electromagnetic spect

1 kilometer (km) = 1 000 m

1 millimeter (mm) = 0.001 m

1 micrometer $(\mu m) = 0.000 001 m$ 1 nanometer (nm) = 0.000 000 001 m

1 picometer (pm) = 0.000 000 000 001 m

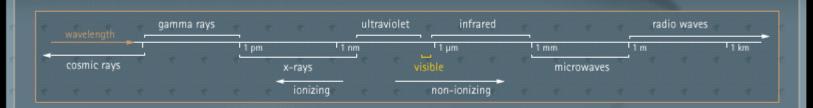




With respect to the wavelength, the electromagnetic spectrum is divided into several parts.

The shortest wavelengths are called [cosmic rays] and [gamma rays], followed by [X-rays] and [ultraviolet light]. Then there is a very small region called [visible light], next to the [infrared] part of the spectrum. [Microwaves] and [radio waves] represent the longest wavelengths of the spectrum.

Lasers usually emit optical radiation that is ultraviolet, visible or infrared radiant energy, with few [exceptions].



With respect to the wavelength, the electromagnetic spectrum is divided into several parts.

λ < 0.01 pm

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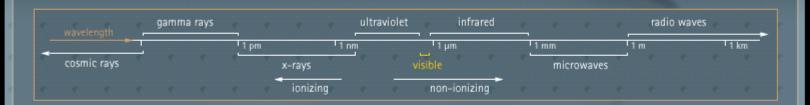


With respect to the wavelength, the electromagnetic spectrum is divided into sev

 $\lambda = 1 \text{ pm} - 10 \text{ nm}$

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The shortest wavelengths are called [cosmic rays] and [gamma rays], for Then there is a very small region called [visible light], next to the [infrarand [radio waves] represent the longest wavelengths of the spectrum.

Lasers usually emit optical radiation that is ultraviolet, visible or infrare

 λ = 100 nm – 400 nm In photobiology, ultraviolet radiation is subdivided into 3 parts:

UV-C 100 - 280 nm UV-B 280 - 315 nm UV-A 315 - 400 nm



With respect to the wavelength, the electromagnetic spectrum is divided into several parts.

s of the spectrum.

Then ther and radio

λ = 400 nm – 700 nm

Visible light is by far the smallest part of the electromagnetic spectrum.

et, visible or infrared radiant energy, with few [exceptions].



With respect to the wavelength, the electromagnetic spectrum is divided into several parts.

The shortest wavelengths
Then there is a very small i
and [radio waves] represe

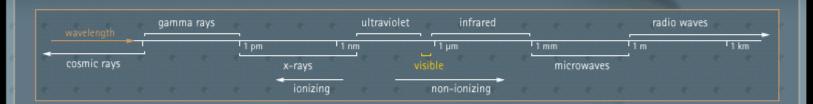
Lasers usually emit optica

λ = 700 nm – 1 mm
In photobiology, infrared radiation is subdivided into 3 parts:

IR-A 700 nm – 1400 nm

IR-B 1.4 μm - 3 μm IR-C 3 μm – 1 mm ved by [X-rays] and [ultraviolet light]. part of the spectrum. [Microwaves]

idiant energy, with few [exceptions].



With respect to the wavelength, the electromagnetic spectrum is divided into several parts.

The shortest wavelengths are called [cosmic rays] and [gamma rays], followed by [X-rays] and [ultraviolet light]. Then there is a very small region called [visible light], next to the [infrared] part of the spectrum. [Microwaves] and [radio waves] represent the longest wavelengths of the spectrum.

Lasers usually emit optical radiation that is ultra-

Research groups sometimes use lasers at other wavelengths, e.g. soft X-ray lasers. At present, those systems are of little relevance for medical applications.

Find the corresponding wavelengths !

[10600 nm] Correct:

10.6 μm is equivalent to...

[1060 nm]

[1.06 mm]

 $[10.64 \ \mu\text{m}]$ 1064 nm is equivalent to... $[1.064 \ \mu\text{m}]$

[0.001064 mm]

Find the corresponding wavelengths!

10.6 μm is equivalent to...

[10600 nm]

[1060 nm]

1064 nm is equivalent to...

[10.64 µm] correct!
[1.064 µm]

Find the corresponding wavelengths!

 $[10600 \ nm]$ 10.6 μm is equivalent to... [1060 nm]

[1.06 mm]

1064 nm is equivalent to...

[10.64 µm] [1.064 µm]

[0.001064 mn correct!

PREVIOUS

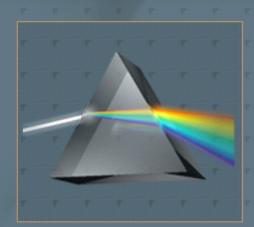
NEXT

Different colors correspond to different wavelengths.

Natural light consists of electromagnetic waves of different wavelengths.

Different wavelengths of the visible part of the spectrum are interpreted as different colors by the human eye: short wavelengths appear blue, longer wavelengths appear red.

The human eye detects different [colors] with different [sensitivities].



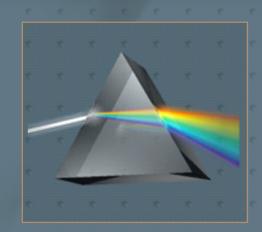
Different colors correspond to different wavelengths.

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Looking at electromagnetic radiation, different wavelengths correspond to different colors. Listening to sound waves, different wavelengths correspond to different pitch. While the human ear can only detect specific sounds the human eye can only detect specific colors. Ultrasound is not audible, ultraviolet radiation is not visible.

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Natural light consists of electromagnetic waves of different wavelengths.

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The

The human eye is most sensitive at a wavelength of $\lambda=550$ nm. Thus, green and yellow light are detected especially well, other wavelengths not as well.

Assuming a green ($\lambda=550\,\text{nm}$) and a red ($\lambda=650\,\text{nm}$) light source appear equally bright to the human eye, this means that in reality the red light is much more powerful.

If those light sources emitted the same output power, the red light would appear much fainter than the green light.

